

Cl 5 Sample Questions With Solution

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When a 1.50 gram sample of pure NOCl is heated at 350 o C in a volume of 1.00 liter, the percent dissociation is found to be 57.2%. Calculate K c for the reaction as written. $\text{NOCl(g)} \rightleftharpoons \text{NO(g)} + 1/2 \text{Cl}_2 \text{(g)}$

Sample Questions - Chapter 17 - Texas A&M University

These sample questions are illustrative of the types of questions used in the Step 3 examination. Although the questions exemplify content on the examination, they may not reflect the content coverage on ... Potassium (K+) 3.5–5.0 mEq/L 3.5–5.0 mmol/L Chloride (Cl ...

Step 3 Sample Questions November 2020 - USMLE

5. A buffer was prepared by mixing 1.00 mole of ammonia and 1.00 mole of ammonium chloride to form an aqueous solution with a total volume of 1.00 liter. To 500 mL of this solution was added 30.0 mL of 1.00 M NaOH. What is the pH of this solution? (a) 8.96 (b) 9.83 (c) 9.31 (d) 9.11 (e) 9.57 6.

Sample Questions - Chapter 19

Suppose that the volume of a particular sample of Cl₂ gas is 8.80 L at 885 torr and 24 degrees C. a. H... View Answer The volume of a gas sample is 400 mL at 25 degrees Celsius and 2.10 atm.

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