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10.17. Out of  $C_6H_5CH_2Cl$  and  $C_6H_5CHClC_6H_5$  which is more easily hydrolysed by aqueous KOH.

Ans:  $C_6H_5CH_2Cl$  is 1<sup>o</sup> aryl halide while  $C_6H_5$

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$\text{CH}(\text{Cl})\text{C}_6\text{H}_5$  is a  $2^\circ$  aryl halide. In  $\text{S}_\text{N}1$  reactions, the reactivity depends upon the stability of carbocations.

Hybridization of Be and Cl in  $\text{BeCl}_2$  - BYJUS  
Class 7 English Sample Papers; ICSE Class 6 Sample Papers. Class 6 Maths Sample Papers; Class 6 Chemistry Sample Papers; Class 6 Physics Sample Papers; Class 6 Biology Sample Papers; ... For example:  $\text{C}_2\text{H}_4 + \text{Cl}_2 \rightarrow \text{C}_2\text{H}_4\text{Cl}_2$ . Ethene + chlorine  $\rightarrow$  1,2-dichloroethane or ethylene dichloride.  
7. Write the balanced chemical equations for the ...

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Sample ...

What is the Hybridization of Beryllium Dichloride? To know about the hybridization of  $\text{BeCl}_2$  (Beryllium Dichloride) we have to take a closer look at the central atom which is Be. Its electronic configuration is  $1s^2, 2s^2$ , where two electrons are present in the valence shell. During the formation of  $\text{BeCl}_2$ , beryllium atom bonds with two chlorine atoms via single covalent bonds.

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