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Since a sulfur atom ( $A_r = 32$ ) weighs twice as much as an oxygen atom ( $A_r = 16$ ), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms. Answer: 32 g 9.2 (a) 1 mole of H atoms = 1 g 1 mole of  $H_2$  molecules =  $2 \times 1 = 2$  g. (b) 1 mole of N atoms = 14 g**

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