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(a) (i) Outline the steps involved
in the purification of water for
town supply. (ii) Give two cations

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that can cause hardness in water.

- (iii) State two disadvantages of hardness of water. (iv) List two sources of water pollution. (b) (i) Explain briefly why water is a good solvent for sodium chloride but not for oil.

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1. A is a solution of hydrochloric acid. B is a solution containing 2.45 g of anhydrous sodium trioxocarbonate (IV) in 250 cm³ of solution. (a) Put A into the burette and titrate it against 20.0 cm³ or

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25.0cm³ portion of B using methyl orange as indicator.Repeat the exercise to obtain consistent titres.

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of CH_4 is. 2. The crystal layers in
graphite are held together by. 3.
Hydrogen bonds are formed
between molecules containing
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The two elements 11X and 19Y

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are in the same group because they have the same.

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