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Unit 7 Chemical Equations and Reactions What is a Chemical Equation? A Chemical Equation is a written representation of the process that occurs in a chemical reaction. A chemical equation is written with the Reactants on the left side of an arrow and the Products of the chemical reaction on the right side of the equation.

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Worksheet 1 Balance the following equations by inserting the proper coefficients. You must keep track of the total number of elements on each side of the equation by keeping a tally beneath each reaction.

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7.3: The Chemical Equation - Chemistry LibreTexts

Unit 7 Chemical Reactions and Equations What is a chemical reaction? • A chemical reaction is a process in which one or more substances are converted into new substances with different physical and chemical properties. – A substance that enters into a chemical reaction is called a reactant. – A substance that is produced by a chemical reaction is called a product.

Date Pd Chemistry Unit 7 Chemical Reactions

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Unit 7: Chemical Reactions I can identify the 5 major types of reactions including: Synthesis, Decomposition, Combustion, Single and Double displacement I can complete a chemical reaction using the activity series and solubility rules

Unit 6: Chemical Reactions

A chemical equation is the symbolic representation of a chemical reaction in the form of symbols and formulae, wherein the reactant entities are given on the left-hand side and the product entities on the right-hand side. The coefficients next to the symbols and formulae of entities are the absolute values of the stoichiometric numbers. The first chemical equation was diagrammed by Jean Beguin ...

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7.05b WS3 Key - Chemistry I Name_KEY Unit 7 WS 3 Yet More

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Piersa, Amanda / Unit 7: Formulas, Equations and Chemical ...
Equation 7.3.1 and Equation 7.3.2 are balanced chemical equations. What is different on each side of the equation is how the atoms are arranged to make molecules or ions. A chemical reaction represents a change in the distribution of atoms but not in the number of atoms.

Unit 7 Chemical Reactions and Equations

Chemistry I Name _____KEY_____ Unit 7 WS 3 Yet More
Balancing Equations Write balanced chemical equations for the following reactions. 1. Ammonia (NH_3) reacts with aqueous hydrogen chloride to form ammonium chloride. $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$ 2. Calcium carbonate decomposes upon heating to form calcium oxide and carbon dioxide. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ 3.

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Unit 7 Chemical Equations and Reactions

Unit 7- Chemical Equations Day Page # Description IC/HW Due Date Completed ALL 2 Warm-up IC 1 3 - 4 Balancing Equations Notes IC 1 5 - 6 Balancing Chemical Equations Worksheet HW 1 Chemical Reactions Pre-Lab HW 2 Types of Chemical Reactions

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Lab IC 3 7 – 8 Type of Chemical Reaction Notes IC

Chapter 7.3: Chemical Equations - Chemistry LibreTexts

Unit 6: Chemical Reactions (Chapter 11) 2 . 3 Reactions: The Chemistry of Change ... 7 Balancing Chemical Reactions

Consider this reaction: $2\text{H}_2 + \text{O}_2 \dots$ Write a balanced chemical equation for the following reactions. d.) Bromine combines with nitrogen to produce nitrogen tribromide.

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