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...

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Because the question involves mass, we will need to know the molar mass of  $C_6H_{12}O_6$ . Using a periodic table we find the molar mass of  $C_6H_{12}O_6$  to be  $180.1 \text{ g}\cdot\text{mol}^{-1}$ .

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CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Formation of solutions where the process is endothermic can be spontaneous provided that \_\_\_\_\_. A) the solvent is a gas and the solute is a solid

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