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• The periodic table is provided for reference on page 2 of this examination paper. • The maximum mark for this examination paper is [30 marks] . 45 minutes

Resources for IB Chemistry SL/HL, 2017 - Mrs. Hala Mattar

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Resources for IB Chemistry SL/HL, 2017, taught by Mrs. Hala Mattar at ACS Hillingdon.

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When 100 cm³ of 1.0 mol dm⁻³ HCl is mixed with 100 cm³ of 1.0 mol dm⁻³ NaOH, the temperature of the resulting solution increases by 5.0 ° C. What will be the temperature change, in C, when 50 cm³

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• Do not open this examination paper until instructed to do so. • Answer all the questions. • For each question, choose the answer you consider to be the best and indicate your choice on the answer sheet provided. • The periodic table is provided for reference on page 2 of this examination paper.

Chemistry Standard level Paper 1 - Xtreme

2209-6110 - 3 - Turn over M09/4/CHEMI/SPM/ENG/TZ1/XX+ 1. The molar mass of a compound is approximately 56 g mol^{-1} . Which formula is possible for this

Chemistry IB SL Paper 1, 2, 3 May 2014? (TZ1/2) : IBO

Paper water (in cellulose fibres of paper); Do not accept just "paper" or "liquid". Clarification: Award M2 is for the "stationary phase" in paper chromatography.

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Exam Discussion: Chemistry SL paper 3 TZ1 : IBO

$\times \times ? 1 \times 1000$ 17. Under which conditions will the reaction between 1.0 g calcium carbonate and excess hydrochloric acid be the fastest? Assume that all reactions are carried out at the same temperature. A. One large piece of calcium carbonate and 2 mol dm^{-3} hydrochloric acid B. One large piece of calcium carbonate and 1 mol dm^{-3} ...

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Subject Details: Chemistry SL Paper 2 Markscheme Mark Allocation Candidates are required to answer ALL questions in Section A [30 marks] and ONE question in Section B

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