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2015 Chemistry Semester 2 Final Exam Study Guide 1 | Page Th
2015 Semester 2 Cumulative Final Exam will be around 100
multiple choice questions. A periodic table and calculator
(chemical symbols – no names) will be provided. NO GRAPHING

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CALCULATORS WILL BE ALLOWED. Resources for studying
(1)Metacognitive Logs (2)Old Quizzes (3)Homework and
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Final Study Guide. Monatomic Ion. Ions formed from a single
atom. Binary Compound. Compounds composed of two elements
Nomenclature. Naming system. Salt. An ionic compound
composed of a cation and the anion from an anion from an acid

Study Guide for Final Exam - Chemistry
Chemistry Semester 2 Final Exam Study Guide Learning goals
Checklist As you prepare for your final exam use the following I

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to monitor your progress. Only check off a learning goal when you are confident of your ability to complete the task described.

Unit 0: Being a Chemist Recognize and apply safe practices when working in the lab.

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CALCULATOR THE DAY OF THE EXAM! You will be provided with a periodic table with the polyatomic ion chart and equations that

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Study Guide - Semester 2 Note: This is a simply a guide of questions to study for the test. It is not a promise that every question on the test is covered here.

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General Chemistry Semester 1 Study Guide 2010 6 c. tin(II)iodide
Sn(IO₃)₂ d. calcium nitrite Ca(NO₂)₂ 16. Name these ternary compounds: a. SrClO₃ strontium chlorate b. AgClO₃

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silver(I)chlorate c. NH_4NO_2 ammonium nitrite d. $\text{Cu}_3(\text{PO}_4)_2$ copper(II)phosphate Unit 6... Covalent Compounds 1. Name the following covalent compounds.

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Electrochemistry) Practice Problems (Skills ...

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Unit 1...Measurement & Classification of Matter

FALL SEMESTER ? ? Honors Chemistry Fall Final Exam

Review: 1. Revisit all old practice tests 2. Revisit all old study guides 3. Do Mrs. Rulis's awesome fall semester study guide found in the back of the chapter 8-9 study guide.

Honors Chem Final Exam Review - Mr. Grosser's Science ...

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Chemistry / Study Guide - Semester 2

Electrolyte an ionic compound whose aqueous solution conducts electric current. Formula unit the simplest ratio of ions represented in an ionic compound. Aqueous solution a solution in which the solvent is water. Chemical equation a statement using chemical formulas to describe the identities and relative amounts of the reactants and products involved in the [...]

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Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 2 For Multiplication or Division: Answer is rounded to the LOWEST # OF SD'S in the data. For Addition

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or Subtraction: Answer is rounded to the LOWEST # OF DECIMAL PLACES in the data. ... Microsoft Word - Study Guide for Final Exam.doc

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