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The Henry's law constant for the solubility of N₂ gas in water at 298 K is 1.0×10^5 atm. The mole fraction of N₂ in air is 0.8. The number of moles of N₂ from air dissolved in 10 moles of water at 298 K and 5 atm pressure is [IIT JEE 2009]

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