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Questions 5 and 6 are based on the following text: Isotopes. The nucleus of an atom contains both protons and neutrons. Protons have a single positive electric charge, while neutrons have a charge of zero. The number of protons that a nucleus contains, called the atomic number and abbreviated as Z, determines the identity of an atom of matter.

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The sample questions are arranged by module. Examples of both objective-response questions and short-answer questions for each of the modules, Equilibrium and Acid Reactions, Acid/base Reactions, Organic Chemistry and Applying Chemical Ideas, are provided. Each sample question has been mapped to show how the question relates to content,

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