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following  
balanced  
chemical equa-  
tions in terms  
of particles,  
moles, and mass.  
Show that the  
law of*

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molecular  
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mass of  
empirical  
formula 198.22  
g/mol/99.11  
g/mol 2(KCO<sub>3</sub>)<sub>n</sub>  
The molecular  
formula of the  
compound is K<sub>2</sub>C  
20 6. Section  
10.5 The Formula  
for a Hydrate 1.

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hydration

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Green leaves

turn yellow and

red as the

seasons change.

Baking powder

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*produces a gas  
that makes a  
cake rise. no In  
your textbook,  
read about how  
to represent  
chemical  
reactions and  
how to balance  
chemical  
equations. Use  
the terms below  
to complete the  
passage. Each*

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*term may be used  
once, more than  
once, or not at  
all. arrow  
reactant plus  
sign product*

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The carbon cycle  
is the process  
by which carbon  
moves from the*

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atmosphere into  
the Earth and  
its organisms  
and then back  
again Chemistry  
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study guide  
answer key. Gain  
a deeper  
understanding of  
how the carbon  
cycle works in  
this . . .

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*Assessment ...*

*Elements from*

*group 18 are*

*relatively*

*unreactive, why.*

*Group 18 are*

*noble gases with*

*complete outer*

*rings and do not*

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*easily form  
ions. Elements  
from group 17  
are reactive  
why. These  
elements only  
need to gain a  
single electro  
to form an  
octet.*

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Guide TE - Mr.  
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States of

Matter. Solid.

Liquid. Gas.

Solid, liquid,

gas, plasma. A

form of matter

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shape and  
volume.

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life.*

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*Technique 4: How*

*it is done:*

*Sequence the*

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*steps of  
separating a  
mixture of sand,  
salt, and iron  
filings.*

*Identify which  
physical  
property you  
were using in  
each step. Mix  
the sand and  
salt mixture  
with water.*

*Physical*

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property used:

Boil the salt

and water

mixture, leaving

the salt behind.

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*Design a concept map that shows the conversion factors needed to convert among mass, moles, and number of particles. Moles  
Moles Number of particles Mass  
1. 2. 3. 4.*

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Matter and  
Change MIXED  
REVIEW SHORT  
ANSWER Answer

*the following  
questions in the  
space provided.*

*1. Classify each  
of the following  
as a homogeneous  
or heterogeneous  
substance.*

*homogeneous a.  
sugar*

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*homogeneous d.*

*plastic wrap*

*homogeneous b.*

*iron filings*

*heterogeneous e.*

*cement sidewalk*

*heterogeneous c.*

*granola bar 2.*

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Thinking

Critically

Answer the

following

questions.

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Change Answer  
Key Chapter 8

5 Chapter  
Assessment 1. A  
radio station  
has a frequency  
of 103.7 MHz. (1  
MHz  $10^6$  s<sup>-1</sup>)  
What is the  
wavelength of  
the radiation  
emitted by the  
station?  
Indicate where  
this ...

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Chapter 18



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