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Formula Review 1. Write the formulas for the following compounds: 2. Write the names for the following ...

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learned about single replacement reactions in which a single atom replaces an ion from another reactant. Study what happens in the following reactions. They are called double replacement reactions.

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10 7.25 10 23 24 x g mol g mol 529.9 44.0 12.04 x g mol g
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c) Using your answers to 15b and 17b, what can be said about elements in the same column and the force of attraction between their outer energy level and nucleus. Elements in the same column have the same attraction between the nucleus and outer level. electrons. 18. Atom A is larger than atom B.

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