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*ANSWER KEY ELEMENT SYMBOLS Name An element symbol can Stand tor one atom of the element or one rnole of atoms of the element. (One mole  $6.02 \times 10^2$ '*

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atoms or on element.) ... Page 27 Chemistry IF8766  
Atom!c Mass 1.007 ,oonq 350 s 2 . os z 3%.oz 5Æ6  
55Fe, 85% 99% H, 0.8% H, 0.2

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Directions: Determine the number of moles in each of  
the quantities below. 1) 52 g of NaCl. ANSWER: 0.89  
mole Chemistry IF8766. Page 50.*

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*Gram Formula Mass Instructional Fair Answer Key  
COMPOSITION OF HYDRATES Name A hydrate is an ionic compound with water molecules loosely bonded to its crystal structure. The water is in a specific ratio to each formula unit of the salt.*

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*SIGNIFICANT FIGURES Name*

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*Stoichiometry: Limiting Reagent 1.  $N_2 + 3H_2 \rightarrow 2NH_3$*

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3 How many grams of  $\text{NH}_3$  can be produced from the reaction of 28g of  $\text{N}_2$  and 25 g of  $\text{H}_2$ ? 2. How many grams of the excess reagent in Problem 1 is left over? 3.  $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$  What volume of hydrogen at STP is produced from the reaction of 50.0g of Mg and the equivalent of 75.0g ...

pg 61 - Predicting Products of Chemical Reactions  
Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name \_ ...  
Page 39 Chemistry IF8766 Page 38 TYPES OF CHEMICAL BONDS Name-----Classify the following compounds (metal + metal), covalent (nonmetal + ...  
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*Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate (NH<sub>4</sub>)<sub>3</sub> PO<sub>4</sub> \_\_\_\_\_ 2. iron (II) oxide*

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*FeO\_\_\_\_\_ 3. iron (III) oxide Fe<sub>2</sub>O<sub>3</sub> ... Chemistry  
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*Stoichiometry: Mole-Mole Problems - Mr. V's  
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*I The answer is expressed as 255.5 mL since 46.0 mL  
has only one decimal place. 11 Perform the following  
operations expressing the answer in the correct  
number of significant figures. 1, 1.35 m x 2.467 m =  
Chem\* IF8766 10 OInstructional Fair, Inc. 60*

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Word Equations Answer Key 1. Zinc and Lead (II)*

