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r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name \_ Ionicbondingoccurswhenameitd hcnstersonormoreelectrons10anonmelolln on ...

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Chemistry If8766 9 QInstructional Fair, Inc. 59. CALCULATIONS USING SIGNIFICANT FIGURES Name ? When multiplying and dividing, limit and round to the least number of significant figures in any of the factors. 11 Example 1 : 23.0 cm x 432 cm x 19 cm = 188,784 cm3 11

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I l . Sugar dissolves in water. 12. Wood rotting 13. Pancakes cooking on a griddle 14. Grass growing in a lawn 15. A tire is inflated with air. 16. Food is digested in the stomach. 17. Water is absorbed by a paper towel, Inc. Chemistry If8766 19 ©Instructional Fair, Inc. 13.

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NAMING HYDROCARBONS Name  
IDEAL GAS LAW Use the ideal Gas Law below to solve the following problems. pressure in atmospheres volume in liters number of moles L atm Universal Gas Constant = 0.0821

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The Mole And Volume Worksheet Instructional Fair Inc  
acid + sodium hydroxide Chemistry If8766 Page 61 Instructional Fair, Inc. Gas Law Worksheet Answer Details 1. Determining Molecular Formulas (True Formulas) reagent in analytical chemistry, dimethylglyoxime, has the empirical This packet is a cumulative review of many topics from the year that are

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KMBT 654-20140114144911 - Berger's Chemistry Class  
Instructional fair inc chemistry if8766 ideal gas law Solve the following problems using the ideal gas law. 1) How many moles of gas does it take. What volume is Indicated on each of lthe maduated

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NAMING HYDROCARBONS Name Name the compounds below according to the IUPAC naming system 1. H H H I I I H —C —C —H I I I H H H 5.

The Mole and Volume - Ms. Agostine's Chemistry Page  
enstructional Fair, Inc. 5. 35 g of CuSOa.5H2O Determine the number of grams in each of the quantities below. 1. 2.5 moles of NaCl 2. 0.50 moles of 3. 1.70 moles of KMnO4 4. 0.25 moles of KCl e, ST 5. 3.2 moles of CuS0405H20 Chemistry If8766

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