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Answer Questions, Exercises, and Problems 29–40. Check your answers with those at the end of the chapter. Workbook If your instructor recommends the Active Learning Workbook, do Questions, Exercises, and Problems 29–40. Chapter 22–Assignment D: Manufactured Polymers

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Quiz: Hydrocarbons

Hydrocarbons An infinite variety of compounds can be assembled from only carbon and hydrogen atoms. Such hydrocarbons are the simplest organic compounds, but they are also of prime economic importance because they include the constituents of petroleum and natural gas.

Hydrocarbon Quizzes Online, Trivia, Questions
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hydrocarbons are used as solvents for paints. They are also used as the starting materials for manufacture of many dyes and drugs. Thus, you can well understand the importance of hydrocarbons in your daily life. In this unit, you will learn more about hydrocarbons.

13.1 CLASSIFICATION Hydrocarbons are of different types. Depending upon the

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132 Chemistry: Matter and Change • Chapter 22
Study Guide for Content Mastery In your textbook, read about natural sources of hydrocarbons and rating gasolines. In the

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space at the left, write the word or phrase in parentheses that correctly completes the statement. 11. (Petroleum, Natural gas) is a mixture of alkanes, aromatic

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Study Guide Key Concepts 22.1 Hydrocarbons • Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar

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molecules. 22.2 Unsaturated Hydrocarbons

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Most hydrocarbons are nonpolar because of the close electronegativities of the C and H atoms. As such, they dissolve only sparingly in H₂O and other polar solvents. Small hydrocarbons, such as methane and ethane, are gases at room temperature, while larger hydrocarbons, such as hexane and octane, are liquids.

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STUDY GUIDE In your textbook, read about organic chemistry, hydrocarbons, and straight-chain alkanes. Use each of the terms below

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just once to complete the passage. homologous series straight-chain alkanes If all of the carbon atoms are linked by single covalent bonds and there are no branches, the compounds are called (1) examples.

SECTION 22.1 HYDROCARBONS(pages 693–701)
2 A Self-Study Guide to the Principles of Organic Chemistry 6.5 Electron Configuration of the Elements Commonly Involved in Organic Compounds Figure 1.1 Hydrocarbon, methane and its derivatives The Periodic Table contains more than one hundred elements; however, organic chemistry deals mostly with only

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certain elements.

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Modern Chemistry 181 Organic Chemistry
CHAPTER 22 REVIEW Organic Chemistry SECTION 2
SHORT ANSWER Answer the following questions
in the space provided.

____1.____Hydrocarbons that contain only
single covalent bonds between carbon atoms
are called (a) alkanes. (c) alkynes. (b)
alkenes. (d) unsaturated.

Chapter 21: Hydrocarbons

It is the study of the chemistry of carbon

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compounds. They contain any number of carbon atoms, one after another, in a chain.

Chapter 22

Hydrocarbon Quizzes & Trivia It is the study of the chemistry of hydrogen compounds. ... A hydrocarbon in organic chemistry is an organic compound that has mostly hydrogen and carbon. This quiz will help you find out how much you know about different types of hydrocarbons and alkanes.

CHAPTER 22 REVIEW Organic Chemistry
Hydrocarbons BIG Idea Organic compounds

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called hydrocarbons differ by their types of bonds. 21.1 Introduction to Hydrocarbons ... chemistry, called organic chemistry, is devoted to their study. Recall that carbon is an element in group 14 of the periodic table, as shown in

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