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Choose the correct symbol listed below to describe each of the chemical reactions listed in questions 1–15. Also write a balanced chemical equation for each reaction. S = synthesis SD = single

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Chapter 1

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14 Study Guide for An Introduction to Chemistry Exercise 2.3 - Cations and Anions: Identify each of the following as a cation or an anion, and determine the charge on each. (Obj 22) a magnesium atom with 12 protons and 10 electrons $(+12) + (-10) = +2$ This is a +2 cation. a fluorine atom with 9 protons and 10 electrons

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3 Use with text pages 86 – 94 CHAPTER Chemistry: Concepts and Applications Study Guide, Chapter 39 Use each of the terms below just once to complete the following passage.

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