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*10th Class Chemistry, ch 15, Water as Solvent - Matric Class Chemistry Chapter 15 - Water and Aqueous Systems - 15 Assessment - Page 511: 64 Answer Anions are attracted to the hydrogen atoms of the water molecule, because hydrogen atoms are slightly positive.*

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*water molecules: if the attraction of the water molecules for the positive ion is stronger than the attraction of the negative ions near it in the crystal, the ion leaves the crystal and enters solution. After entering solution, the*

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*dissolved ion is surrounded completely by water molecules, which tends to prevent the ion from reentering the ...*

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P15 shows .1 four lines to describe the  
possible dependence of percent  
ionization of acetic acid with  
concentration. We are to choose the  
one that best represents the trend for  
this weak acid. Analyze*

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