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Lesson Summary. Sublimation is a type of phase transition, or a change in a state of matter, just like melting, freezing, and evaporation. Through sublimation, a substance changes from a solid to a ...

What is Sublimation in Chemistry? - Definition, Process ...

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10.8.1. Anti-Markovnikov addition of HBr to alkenes  
We saw in section 10.4 that under normal conditions, HBr adds to an unsymmetrical alkene to form an alkyl halide where the H goes onto the less substituted carbon, and the Br goes onto the more substituted carbon – thus, it obeys Markovnikov's Rule. However, when heated in the presence of a dialkyl peroxide (often written as ROOR), a ...

10.8. Anti-Markovnikov additions to alkenes and alkynes ...

Introduction. The electron configuration is the standard notation used to describe the electronic structure of an atom. Under the orbital approximation, we let each electron occupy an orbital, which can be solved by a single wavefunction. In doing so, we obtain three quantum numbers ( $n, l, m_l$ ), which are the same as the ones obtained from solving the Schrodinger's equation for Bohr's hydrogen atom.

Electron Configuration - Chemistry LibreTexts  
Answers. 2 S. 5 D. Under jj coupling the spin quantum number is not a "good" quantum number any more, which means it cannot describe and differentiate electronic states properly. A certain term symbol represents a certain value of L as well as of S, which implies limited possible values of  $M_L$  and  $M_S$ .

Atomic Term Symbols - Chemistry LibreTexts  
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