

## Chemistry And Ph Answers

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Cabbage Chemistry | Science Project

(a) A research-grade pH meter used in a laboratory can have a resolution of 0.001 pH units, an accuracy of  $\pm 0.002$  pH units, and may cost in excess of \$1000. (b) A portable pH meter has lower resolution (0.01 pH units), lower accuracy ( $\pm 0.2$  pH units), and a far lower price tag. (credit b: modification of work by Jacopo Werther)

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Because red cabbage has this indicator pigment, it is possible to determine the pH of a solution based on the color it turns the red cabbage juice. The pH of a solution is a numerical measure of how basic or acidic it is. A solution with a pH between 5 and 7 is neutral, 8 or higher is a base, and 4 or lower is an acid.

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