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In 100 atoms of boron there are: 81 atoms of mass 11 = $81 \times 11 = 891$ 19 atoms of mass 10 = $19 \times 10 = 190$ Total mass of 100 atoms = 1081

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Chemistry Live! – Worked Solutions 2 Chapter 9 - The Mole Concept 9.1 Since a sulfur atom ($A_r = 32$) weighs twice as much as an oxygen atom ($A_r = 16$), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms.

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