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Workbook Answer

Key 14 4

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Chapter 11.2

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KEY 1. Watch the

video and then

complete the

chart. Teacher

notes on next

page! Colors: A

= Red, B = Blue,

C = Green, D =

Yellow. 2. Use

colored pencils

to circle the

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Key 1114

common atoms or  
compounds on  
each side of the  
equations to  
help you  
determine the  
type of reaction  
it illustrates.

Chapter 11

Chemical

Reactions Answer

Key Chapter 11

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Workbook Answer

Key 14

reactions. A chemical reaction is a process that changes one set of chemicals into another set of chemicals.

The elements or compounds that enter into the reaction are the reactants. The elements or

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Worksheet Answer

Key 14  
compounds  
produced by the  
reaction are the  
products.

Chemical

reactions

involve changes

in the chemical

bonds that join

atoms in

compounds.

Energy in

Reactions Some

chemical

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Reactions

reactions

release energy;

others absorb

energy.

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2.4 Chemical

Reactions and

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Physical Science

Reading and

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Chapter 7 73

Section 7.1

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Key (192-198)

This section

discusses the use of chemical equations and how to balance them. It also

demonstrates the use of calculations in chemistry.

Reading Strategy

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Reactions

(page 192)

Workbook Answer

Monitoring Your

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Preview the Key

Concepts, topic

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combination

reaction. a

chemical change

in which two or

more substances

react to form a

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Reactions

substance.

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reaction. a

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in which a

single compound

breaks down into

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products; only

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Section 3

Answers CHAPTER

11 CHEMICAL

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Chemical

reactions

involve the

breaking of

chemical bonds

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and the single

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Chapter 11

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Reactions

Worksheet Answer

Key 144

Chemical  
Reactions Guided  
Reading And  
Study ...

A chemical  
reaction is a  
process that  
changes one set  
of chemicals  
into another set  
of chemicals. ?

The elements or  
compounds that  
enter into the

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Reactions

Workbook Answer

Key 444

reaction are the  
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REVIEW SHORT

ANSWER Answer

the following

questions in the

space provided.

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1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products. (c) identities of

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Key 11.1

reaction.

11.2 Types of

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Reactions >

A Workbook for

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Equilibria ...

if the workbook

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Key 14.1

for students to learn the material who might not have learned it otherwise.

Hence, ... 14.1

Reaction

Coordinate The

key point of this section is for you to review mole balances and



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their

implications.

You should be

familiar with

11.2 Types of

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Henry County

School ...

The bag bubbled,

got cold, and

turned from blue

to red. Thus,

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Reactions

Workbook Answer

Key 144

the student  
determined that  
it was a

chemical  
reaction. Then,  
they responded  
to the first  
question of the  
lab station.

Take a look at  
your lab  
document -  
notice that the  
questions are

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Chemical  
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Answer Key.

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Title Cards ...

2.4 Chemical

Reactions and

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Combination

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The first type of reaction is the combination, or synthesis, reaction. A combination reaction is a chemical change in which two or more substances react to form a single new substance. As shown in Figure

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11.5, magnesium metal and oxygen gas combine to form the compound magnesium oxide.

Chemical

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Answer Key -

BetterLesson

1) All chemical reactions are chemical

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## Chemical

### Reactions

#### Workbook Answer

changes. 2) New properties appear. 3) The bonds between the atoms are broken and the atoms recombine 'n new ways. The carbon dioxide gas was produced through the chemical reaction in the bag. Be sure

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Chemical

Reactions

Workbook Answer

Key 111  
students don't  
misinterpret it  
as a change of  
state. Page 4.

Chapter

7 Chemical

Reactions

Section 7.1

Describing

Reactions

A place where

you can ask,

help, and share.

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Chemical

Reactions

CCSS Math.

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Chapter 7:

Chemical

Reactions

Vocabulary/Key

Terms ...

11.2 Types of

Chemical

Reactions > 30 A

halogen can also

replace another



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## Chemical

### Reactions

#### Workbook Answer

#### Key 11

halogen from a compound. •The activity of halogens decreases as you go down Group 7A of the periodic table—fluorine, chlorine, bromine, and iodine. •Bromine is more active than iodine, so this reaction

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