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Chapter 7 - Chemical Formulas &  
Chemical Compounds - yazvac  
Chapter 7 Significance of a Chemical  
Formula, continued □ For an ionic  
compound □ the chemical formula  
represents one formula unit □ the  
simplest ratio of the cations (+) and  
anions (-)  $2 \text{SO}_4$  □ Note also that

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there is no subscript for sulfur: when there is no subscript next to an atom, the subscript is understood to be 1.

Chemical Formula of Common  
Compounds, Chemical reaction ...  
Chapter 7: Chemical Formulas and  
Chemical Compounds Section 7-1:

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Chemical Names and Formulas 7-1-1

Explain the significance of a chemical formula. The subscript indicates that there are 8 carbon atoms in the molecule of octane. The subscript indicates that there are 18 hydrogen atoms in the molecule of octane.

Subscript 2 refers to 2 aluminum ...

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CHAPTER 7 Chemical Formulas and  
Chemical Compounds  
Importance of Chemistry Formulas  
Types of Chemistry Formulas or  
Chemical Formula How to write  
Chemical Formula List of Chemical

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Compound Formula. The chemical formula gives us a clear understanding regarding the chemical substances, like how many and what the chemical elements of the Periodic table consists of, along with the way they are ...

chemistry chemical formulas

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Chemical Compounds

Chapter 6 "Chemical Bonding"

Chapter 7 "Chemical Formulas and  
Chemical Compounds" Chapter 8

"Writing and Balancing Equations"

Chapter 9 "Stoichiometry" Chapter 11

"Gas Laws" Chapter 14 "Lewis Dot  
Structures, Acids & Bases" VSEPR

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Chemical Formulas And Compounds  
Chapter

SECTION 1 continued 6. Some binary

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compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

Chapter 7: Chemical Formulas and  
Chemical Compounds

Chapter 7 □ Chemical Formulas and

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Chemical Compounds 7-1 Chemical Names and Formulas I. Significance of a Chemical Formula A. Molecular formulas 1. Number of atoms of each element in one molecule of a compound C<sub>2</sub>H<sub>6</sub> = ethane (2 carbon atoms, 6 hydrogen atoms) B. Ionic Compounds 1.

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chemical formulas ...

Chapter 7 Significance of a Chemical  
Formula □A chemical formula indicates  
the relative number of atoms of each  
kind in a chemical compound. □For a

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molecular compound, the chemical formula reveals the number of atoms of each element contained in a single molecule of the compound. □ example: octane □ C<sub>8</sub>H<sub>18</sub> The subscript after the C

Chapter 7: Chemical Formulas and

*Page 23/33*

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## Chemical Compounds

Chapter 7 - Chemical Formulas & Chemical Compounds Chapter 7 is the final chapter of the first semester. It serves as a summary of all course content discussed so far, as the focus of this chapter is on process skills rather than new concepts or theory.

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Chapter 7 - Chemical Formulas and  
Compounds Flashcards ...

Chapter 7 Section 1 Chemical Names  
and Formulas Significance of a  
Chemical Formula, continued □The  
chemical formula for an ionic  
compound represents one formula

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unit—the simplest ratio of the compound—s positive ions (cations) and its negative ions (anions).

□example: aluminum sulfate □Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>

## 7 Chemical Formulas and Chemical Compounds

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chemists use systematic methods for naming compounds and for writing chemical formulas. In this chapter, you will be introduced to some of the rules used to identify simple chemical compounds. Significance of a Chemical Formula Recall that a chemical formula indicates the relative

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number of atoms of each kind in a chemical compound.

Chemical Formulas and Compounds  
Chapter 7: Chemical Formulas and  
Chemical Compounds Section 2:  
Oxidation Numbers Oxidation numbers  
An oxidation number, also called

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oxidation state, indicates the general distribution of electrons among the bonded atoms in a molecular compound or polyatomic ion. Unlike ionic charges, oxidation numbers do not have an exact physical meaning.

7 Chemical Formulas and Chemical

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## Compounds

Start studying Chemistry chapter 7: Chemical formulas and chemical compounds,. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Chemical Formulas and Chemical

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## Chapter 7 Compounds

### CHAPTER 7 REVIEW Chemical

### Formulas and Chemical Compounds

### MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b.



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