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Atoms with close
electronegativity values
(like two nonidentical

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nonmetals) form polar covalent bonds. If the electronegativity difference is large (between a metal and a nonmetal) then ionic bonds form. From here, you can test yourself to see if you know the trends in the periodic table or you might wish to review the types of chemical bonds.

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Multiple Choice

Questions (MCQs) -

Quiz ...

Chemical Bonding Quiz

#1. 1. ... All of the

answers are correct D)

Ionic. 5. In ionic

bonding; A) electrons are

shared B) Electrons are

given away C) Electrons

are accepted D) Two

answers are correct. 6. ...

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Chapter 5 Bonding Test -
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Predict the type of bonding in substance A.
(1 pt.) 31) Given the binary compound formed from magnesium and chlorine: a) Write

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the correct IUPAC name
for this compound (1

pt.) b) Write the correct
chemical formula for this
compound (1 pt.) c)

What type of bond forms
between magnesium and
chlorine? [Give one
reason to support your
answer.] (2 pts.)

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Bonding Practice Test

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Multiple Choice Identify
the choice that best
completes the statement
or answers the question.

- _____ 1. A mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together is called a(n) a. dipole. c. chemical bond. b. Lewis structure. d. London force. _____ 2.

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Answers Chemical Bonding Quiz #1

Which of the statements given concerning chemical bonding is FALSE? Energy is required to break any chemical bond. Energy must be released if two atoms are to form a chemical bond. A chemical bond occurs only if the potential

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energy is lowered. All chemical reactions require a transfer of electrons in forming new bonds.

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Press Print for the flash cards! Terms in this set (36) What is Chemical Bonding? The joining of atoms to form a new substance - the properties are different because they formed a new substance. What is a Chemical Bond?

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Chemical Bonds. A water molecule (H_2O) is an example of a covalent bond because _____.

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that can be found at

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To take the quiz, click on

the answer. The circle

next to the answer will

turn yellow. You can

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change your answer if
you want.
Answers

Chemical Bonding Quiz #1

Chemical Bonding
SECTION 1 SHORT
ANSWER Answer the
following questions in the
space provided. 1. a A
chemical bond between
atoms results from the
attraction between the
valence electrons and of

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Answers
different atoms. (a)
nuclei (c) isotopes (b)
inner electrons (d) Lewis
structures 2. b A covalent
bond consists of (a) a
shared electron. (c) two
...

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Multiple Choice Identify

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the choice that best completes the statement or answers the question.

_____ 1. What is the name given to the electrons in the highest occupied energy level of an atom?

... Test Author: Ryan Huerto
Created Date:

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Which of the following is
NOT a type of chemical
bond? A) All of the
answers are correct B)
Ionic C) Covalent D)
Metallic. 5. In ionic
bonding; A) Electrons
are given away B)

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Electrons are accepted
C) Two answers are
correct D) electrons are
shared. 6. ... Created with
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'Next' to see the next set
of questions. You can
skip questions if you
would like and come
back to them ...

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Chapter 5 Bonding Test
A. Do Not Write on This
Test!!! You may use a
periodic table. Multiple

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Choice-Identify the choice that best completes the statement or answers the question. Electrons involved in bonding between atoms are. ... A chemical bond formed when two atoms share electrons is called a(n)

Chemical Bonds in
Compounds Quiz
Modern Chemistry 46

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Chapter Test Chapter:

Chemical Bonding In the

space provided, write the

letter of the term or

phrase that best

completes each statement

or best answers each

question. _____ 1. The

charge on an ion is a.

always positive. b. always

negative. c. either

positive or negative. d.

zero.

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Quizizz. A chemical
bond is: Chemical
Bonding DRAFT. 6th -
10th grade. 1445 times.
Chemistry. 67% average
accuracy. 4 years ago.
mdrucker. 4. Save. Edit.
Edit. Chemical Bonding
DRAFT. ... answer
choices . Made up of

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electrons. An attractive force that holds atoms together. All of the above. None of the above. Tags: Question ...

Assessment Chapter Test A

The millions of different chemical compounds that make up everything on Earth are composed of 118 elements that bond together in

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different ways. This module explores two common types of chemical bonds: covalent and ionic. The module presents chemical bonding on a sliding scale from pure covalent to pure ionic, depending on differences in the electronegativity of the bonding atoms.

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