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industrially by reacting  
nitrogen and

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hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is  $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$ . If 4.0 mol of  $H_2$  react, how many moles of  $NH_3$  will ...

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vapour pressure is  
about 20 mmHg at  
this temperature. Ch4

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police car is searching  
for a fugitive that  
managed to escape a  
while ago. If you look  
at a different metal,  
say Au right next to  
Hg on the Periodic

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The heat produced by  
burning of 1. During

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an endothermic reaction, energy is absorbed into the system and enthalpy increases. However, we can also answer

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Water's saturated vapour pressure is about 20 mmHg at this temperature.  $\text{CH}_4$   $\text{O}_2$   $\text{CO}_2$   $\text{H}_2\text{O}$  balanced equation methane combustion reaction material balances on

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