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ChemActivity 30

Limiting Reagent

Information 177

Chemists refer
to the reactant
which limits the

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amount of product that can be made from a given collection of original reagents as the limiting reagent or limiting reactant.

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(See
chemactivity 5
for explanation)

Two atoms have the same nuclear charge, but one atom has a lower IE than the other atom. How could this happen? The first atom must have a greater

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distance between the valence electron and the nucleus than the other atom.

**different distances change results.

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Myriah Day will post the HOMEWORK KEYS ON THIS PAGE.

PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time.

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The Reaction
Quotient 2294.
Predict which of
the following
will happen to
the moles of
PCl₅ after

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injection of the 0.0600 moles of PCl_3 . a) No change in the moles of PCl_5 because the system is at equilibrium. b) PCl_3 and Cl_2 will be consumed to form more PCl_5 . c) PCl_5 will be consumed to form more

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PCl₃ and Cl₂.

What are the factors of 40? -
Answers

ChemActivity 32
Molarity 4. Draw
a figure similar
to the right
side of Figure 1
for in solution
showing exactly
3 cations and
the appropriate

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number of anions. (You can omit 1-120 molecules for clarity.)

Information The concentration of a solute in an aqueous solution can be expressed in many

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reaching

equilibrium:

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ChemActivity 38:

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Sep 8: Solving
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of the
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constant:

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ChemActivity 40a

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40b:

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...

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4. The bond energy (in MJ/mole) is the energy required to break one mole of the specified bonds.

Triple bonds share six electrons between two atoms and are

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stronger (have greater bond energy) than double bonds, which share four electrons between two atoms.

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The factors of 40 are 1, 2, 4, 5, 8, 10, 20 and 40. The factor pairs of 40 are 1 x 40, 2 x 20, 4 x 10, and 5 x 8. The factors of 40 are 1, 2, 4, 5, 8, 10, 20, and 40. A factor = any number which ...

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answers given

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here are less detailed than what is expected for a student's answer.

ChemActivity 1

1. 6, 6, 6 2. 6, 7, 7 3. 6, 6, 7

4. All carbon atoms and ions have six protons in the nucleus.

5. All hydrogen atoms and ions

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have one proton in the nucleus.

6. Z is the number of protons in the nucleus of that atom. 7.

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