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electrons 3(a) Na 2 O ionic Strong forces
between ions/strong ionic bonding SiO 2
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H or all C in same Do n environment OR
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(a) A solution which resists change in pH when small amounts of acid or base are added / on dilution NH_4Cl (allow other correct strong acid) 1 1 1 Do not allow pH remains constant 5 (b) (i) $K_a = [\text{H}$

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WMP/Jan13/CHEM2 Do not write outside the box 4 (04) 13 1(c) Barium metal reacts very quickly with dilute hydrochloric acid, but it reacts more slowly with water. 1 (c) (i) Write an equation for the reaction of barium with water. (1 mark) 1 (c) (ii) A solution containing barium ions can be

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used to show the presence of sulfate ions in an aqueous solution of sodium sulfate.

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