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9 stoichiometry chapter test b. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Jessica_Sifuentes7. Terms in this set (15) the amount of products that form when a reaction is carried out in the laboratory. actual yield. the calculation of quantities in chemical equation.

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Chapter Test B Chapter: Stoichiometry PART I Inthe space provided, write the letter ofthe term or phrase that best completes each statement or bestanswers each question. 1.

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Knowing the mole ratio of a reactant and product in a chemical reaction would allow you to determine a. the energy released in the reaction. b. the speed of the reaction.

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Modern Chemistry 88 Chapter Test Chapter: States of Matter

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers

each question. _____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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Example 1. How many molecules of SO_3 are needed to react with 144 molecules of Fe_2O_3 given this balanced chemical equation?. $\text{Fe}_2\text{O}_3 (\text{s}) + 3\text{SO}_3 (\text{g}) \rightarrow \text{Fe}_2 (\text{SO}_4)_3$. Solution. We use the balanced chemical equation to construct a conversion factor between Fe_2O_3 and SO_3 . The number of molecules of Fe_2O_3 goes on the bottom of our conversion factor so it cancels with our given amount ...

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