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Combustion Reaction

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Involves Oxygen : energy is released as heat

Chapter 8 Lesson 2 Chemical

Point out the chemical formula for each water molecule, H_2O . Explain the formation of the H_3O^+

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ion. ... Then explain that in Chapter 6, Lesson 9, they will combine acids and bases in an indicator solution with the goal of making the pH of the final solution neutral. Standards.

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10; ... Therefore the
chemical formula = Al_2O_3 .

2. Phosphorus trichloride.

Valencies of elements: P = 3
(Group V) Cl = 1 (Group VII)

Lowest common multiple of
valencies = 3. P = $1 \times 3 =$
3.

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periodic table and atoms in Chapter 4. For now, all students need to focus on is the size and mass of the atoms. Tell students that the difference in density between the small, medium, and large samples that they

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measured can be explained based on the atoms and molecules they are made from.

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Thus, the symbol 2NaHCO_3

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indicates two units of sodium bicarbonate, which contain 2 Na atoms, 2 H atoms, 2 C atoms, and 6 O atoms ($2 \times 3 = 6$, the coefficient times the subscript for O).

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with the poem (Page 35)

Question 1: Find three or
four phrases in stanzas one
and two which are likely to
occur in a geography lesson.

Answer:

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