

Chapter 7 Solutions And Colloids

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7.2 Types of Solutions. In Chapter 1, you were introduced to the concept of a mixture, which is a substance that is composed of two or more substances. Recall that mixtures can be of two types: Homogeneous and Heterogeneous, where homogeneous mixtures combine so intimately that they are observed as a single substance, even though they are not.

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Chapter 11 Solutions and Colloids Figure 11.1 Coral reefs, such as this one at the Palmyra Atoll National Wildlife Refuge, are vital to the ecosystem of earth's oceans but are threatened by climate change and dissolved pollution. Marine life depends on the specific

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Unformatted text preview: CHAPTER 7 Solutions and Colloids :5" SECTION 7.1 Physical States of Solutions 1.A solution is a homogeneous mixture of two or more substances. Although solutions are "mixtures" of more than one substance, they are "homogeneous" meaning they are uniform and look the same throughout. Components in a solution are distributed uniformly and appear homogeneous.

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Colloids. Introduction. Solutions are homogeneous mixtures of two or more substances, in which individual components in the mixtures are indistinguishable from one another. A solution composed of one or more solutes and a solvent - the latter is one that is present in the largest quantity. Aqueous solutions are solutions that have water as ...

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Solutions exhibit completely different behavior from suspensions. A solution may be colored, but it is transparent, the molecules or ions are invisible, and they do not settle out on standing. A group of mixtures called colloids (or colloidal dispersions) exhibit properties intermediate between those of suspensions and

solutions . The particles ...

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The particles are spread evenly throughout the dispersion medium, which can be a solid, liquid, or gas. Because the dispersed particles of a colloid are not as large as those of a suspension, they do not settle out upon standing. The table below summarizes the properties and distinctions between solutions, colloids, and suspensions.

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This video is of CBSE Class 9 Science from Chapter "Is Matter Around Us Pure?" ... Colloids - Solutions in which the size of particles lies in between those of true solutions and suspensions are ...

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Chapter 7: Solutions and Colloids - Homework Practice ANSWERS 1. In a solution, which component is considered the solvent? Which is the solute? 2. Identify the solvent and the solute in the following solutions: a. glass of Kool-aid c. An alcoholic beverage solvent is H₂O; solute is Kool-aid solvent is H₂O; solute is alcohol b. carbonated water d. ...

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Chapter 7 Solutions and Colloids 7.7b and 7.8 Solution Properties and Colloids. 2 Solutions Solutions • contain small particles (ions or molecules). • are transparent. • do not separate. • cannot be filtered. • do not scatter light. 3 Colloids Colloids ...

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