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CHAPTER 6 STUDY
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BONDING

INTRODUCTION TO
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BONDING SHORT

ANSWER Answer the following questions in the space provided. 1.

_____ A chemical bond between atoms results from the attraction between the

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Questions. 1.

Describe how covalent and ionic bonds differ with regard to melting point, boiling point, and bond strength. 2.

How is bond length related to stability and energy? 3. Describe the relationship between: a) the

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*difference of the
electronegativity
values of a
compound's atoms,
and ...*

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Questions. 1.
Describe how
covalent and ionic*

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bonds differ with regard to melting point, boiling point, and bond strength. 2. How is bond length related to stability and energy? 3. Describe the relationship between: a) the difference of the electronegativity values of a compound's atoms, and

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Questions. 1.*

*Describe how
covalent and ionic
bonds differ with
regard to melting
point, boiling point,
and bond strength. 2.*

*How is bond length
related to stability and*

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energy? 3. Describe the relationship between: a) the difference of the electronegativity values of a compound's atoms, and

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*Chapter 6: Chemical
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Subject: Chemistry

*Keywords: covalent,
ionic, bonding,
hydrogen bond,
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CHAPTER 6 REVIEW

Chemical Bonding

Question: Chapter-6

Bond Evaluation 1-

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Bond Paying

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With Par Value Of
\$1,000 Coupon Rate
Of 6% And Maturity
Of 2 Years. Market
Rate Is 8%. 2- A Bond
With Par Value Of
\$1000, Coupon Rate
Of 10%, Is Now
Selling For \$1,079.85,
Yield To Maturity For
This Bond Is 8%.

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Practice Questions
Multiple Choice
Identify the choice
that best completes
the statement or
answers the question.*

 1. What is the
name given to the
electrons in the

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highest occupied energy level of an atom? a. orbital electrons c. anions b. valence electrons d. cations

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about the building
blocks of life. For
each statement
below, write true or*

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- false. true* 1. Carbon atoms can bond together in straight chains, branched chains, or rings. *false*
2. Large molecules containing carbon atoms are called micromolecules. *false*
- 3.

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Modern Chemistry 41

Chemical Bonding

CHAPTER 6 REVIEW

Chemical Bonding

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

_____ A chemical bond between atoms results from the attraction between the valence electrons and _____ of different

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*atoms. (a) nuclei (c)
isotopes (b) inner
electrons (d) Lewis
structures 2.*

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Bond Theory-Central tenants-looks at social bonds and delinquency rates.

This theory explains why people don't commit crimes.

Delinquency occurs when social bonds are weak or absent.

Therefore, social bonds and delinquency are interconnected. This

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and how loss of
control allows one to
...*

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