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nucleus (radioisotope) releases energy by emitting radiation during the process of radioactive decay. Nuclear reactions of a given radioisotope cannot be speed up, slowed down, or turned off.

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Mass number and atomic number are conserved in all nuclear reactions. The mass of a 25.0 g piece of  $^{238}\text{Cm}$  (half-life: 2.4 hr) will be reduced to 147 3.1 g after 7.2 hr. Study Guide for Content Mastery Chemistry: Matter and Change Chapter 25

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nuclear radiation are alpha radiation, beta radiation, and gamma radiation. Table 25.1 summarizes the characteristics of these three types of radiation. The different types of radiation can be separated by an electric

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Field as shown in Figure 25.1. Alpha Radiation  
The type of radiation called alpha radiation consists of

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radioisotope. unstable isotopes (any element  
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True-False 16. ST 17. NT 18. AT 19. NT 20.



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Chapter 25 Nuclear Chemistry Figure 25-2 Both  
Pierre and Marie Curie played important roles  
in founding the field of nuclear chemistry.  
Marie Curie went on to show that unlike  
chemical reactions, radioactivity is not  
affected by changes in physical conditions  
such as temperature and pressure.

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