

Chapter 25 Nuclear Equations Worksheet Answer Key

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Chapter 25

n in equation 5? 8. Write the nuclear symbol for the missing term in equation 5. 9. Write the nuclear symbol for the missing term in equation 6. Date Class Balancing Nuclear Equations Balancing Nuclear Equations MATH SKILLS TRANSPARENCY WORKSHEET Use with Chapter 25, Section 25.2 41

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21.2 Nuclear Equations – Chemistry

The production of energy in a nuclear reactor can be stopped by pulling out all control rods. A breeder reactor produces more fuel than fission products produced in nuclear power plants are not radioactive. An uncontrolled chain reaction led to the nuclear accident in Chernobyl, Ukraine. Chemistry: Matter and Change Chapter 25 149

Chapter 25 – Nuclear Chemistry

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NUCLEAR EQUATIONS WORKSHEET ANSWERS

The subscripts and superscripts are necessary for balancing nuclear equations, but are usually optional in other circumstances. For example, an alpha particle is a helium nucleus (He) with a charge of +2 and a mass number of 4, so it is symbolized ${}^4_2\text{He}$. This works because, in general, the charge and mass number are important in the balancing of nuclear ...

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Nuclear Reactions Review Worksheet

strong nuclear force" between protons or neutrons GOOD This attracts nuclides, holds nucleus together Unknown how the strong nuclear force exists. It's existence and strength is really known by deduction! The neutron/proton ratio increases with larger nuclei. This enables the strong force to increase at a pace that can balance

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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating radiation from a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or particles.

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Nuclear decay with no mass and no charge An electron Least penetrating nuclear decay Most damaging nuclear decay to the human body is alpha decay that can be stopped by skin or paper. 3. Owl phcx bescx 10. Nuclear decay that can be stopped by aluminum. Complete the following nuclear equations. 12. ${}_{14}^{235}\text{Pu} \rightarrow {}_2^4\text{He} + \text{C}$ a ${}_{94}^{239}\text{Pu} \rightarrow {}_2^4\text{He} + {}_{92}^{235}\text{U}$ 11. 13. 15 ...

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NUCLEAR EQUATIONS WORKSHEET ANSWERS 1. Write a nuclear equation for the alpha decay of ${}_{91}^{231}\text{Pa}$. ${}_{91}^{231}\text{Pa} \rightarrow {}_2^4\text{He} + {}_{89}^{227}\text{Ac}$ 2. Write a nuclear equation for the beta decay of ${}_{87}^{223}\text{Fr}$. ${}_{87}^{223}\text{Fr} \rightarrow {}_{88}^{223}\text{Ra} + {}_{-1}^0\text{e}$ 3. Write a nuclear equation for the alpha and beta decay of ${}_{62}^{149}\text{Sm}$. ${}_{62}^{149}\text{Sm} \rightarrow {}_2^4\text{He} + {}_{60}^{145}\text{Pm} + {}_{-1}^0\text{e}$ 4.

Chapter 25 Nuclear Equations Worksheet

Chapter 25 Nuclear Chemistry 25.1 Nuclear Radiation 25.2 Nuclear Transformations 25.3 Fission and Fusion 25.4 Radiation in Your Life
balanced nuclear equation, the sum of the mass numbers (superscripts) on the right must equal the sum on the left. • The same is true for atomic numbers

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