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Chapter 2 Objectives (IWBAT)
The student evaluates the energy needs of the human body and processes through

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those needs 1. Differentiate matter from energy. 2. List four major energy forms, and provide one example of how each energy form is used in the body. 3. Define chemical element, and list the four elements that form the bulk of body matter.

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Proteins are constructed from building blocks called amino acids; 20 common types of amino acids are found in the body. Amino acid sequence determines the proteins constructed (Figure 2.12). Fibrous, or structural, proteins are the

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basic structural materials of the body.

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Chapter 2: Basic Chemistry

Homework Chapter 2-Chemistry

Comes Alive Period: Date:

... pH 8-14 is basic or alkaline with 8,9 weak bases, 12,13,14 strong bases
... Do Chapter 2 review

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packet Do myaandp.com
quizzes especially multiple
choice level 1 . Human blood
normally has a pH between
7.35 and 7.45. By
comparison, saliva has a pH

Quia - Chapter 2: Basic Chemistry

1 Chapter 2.1, 2.2 Review
Packet – Answer Key 2.1 The
Nature of Matter Lesson
Summary Atoms The atom is
the basic unit of matter,
made up of three subatomic
particles. Protons have a
positive charge and neutrons
carry no charge. Strong
forces bind protons and
neutrons together in the
nucleus.

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Chapter 2 Basic Chemistry - Welcome to Coach Armstrong's class

Chapter 2: Chemistry Comes
Alive Basic Chemistry

Chapter 2: Chemistry Comes
Alive Matter The “stuff” of
the universe. Anything that
occupies space and has mass.

States of matter Solid

Liquid Gas Basic Chemistry

Chapter 2: Chemistry Comes
Alive Energy Less tangible
no mass, does not take up
space, & is only measured by
its effects on matter.

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Section Review 2-2 1. a 2. b
3. c 4. c 5. a 6. b 7. c 8.
Polarity in a water molecule
is caused by an uneven
distribution of electrons
between the oxygen and
hydrogen atoms. 9. The
concentration of H ions
determines whether a
solution is acidic or basic.
10. Capillary action is the
effect of water rising in a
narrow tube against the
force of ...

Chapter 2 - Basic Chemistry - SlideShare

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Help: Chapter 2 Anatomy I

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Basic Chemistry Concepts ...

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Mrs. Niemoth / Chapter 2 Basic Chemistry

Chapter 2 - Basic Chemistry
1. The Chemistry of
LifeChapter 2 2. Chemistry
All living things depend on
chemistry The structure and
function of all living
things are governed by the

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laws of chemistry Why is it important to understand chemistry in biology? Biologists study chemistry because all living things are made of the same kinds of matter that make up nonliving things

Ch. 2 Answer Key - Lawndale High School

CHAPTER 2 Basic Chemistry.
BASIC CHEM REVIEW

Differentiate between matter and energy. matter is anything that takes up space and has mass energy is the capacity to do work or put matter into motion

Biology Chapter 2: Basic Chemistry | Chemistry

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Flashcards ...

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CHAPTER 2 Basic

Chemistry.doc |

Carbohydrates | Atoms

Chapter 2 Basic Chemistry;
Chapter 20 Lymphatic System;
Chapter 21 Immune System;
Chapter 23 Digestive System;
Chapter 24 Urinary System;
Chapter 26 Reproductive
Systems; Chapter 27 Human
Development and Heredity;
Chapter 3 Cells; Chapter 4
Cell Metabolism; Chapter 5
Microbiology Basics; Chapter
6 Tissues and Membranes

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**chapter_2.1 - Chapter 2.1
2.2 Review Packet Answer Key
2.1 ...**

chapter 2 basic chemistry
A&P worksheet. Terms in this
set (23) Acids Bases and
Salts. substances that
ionize in water; good
electrolytes. Base. Proton
(H⁺) acceptor. Salts and
Buffers. ionize in water to
release hydrogen ions and a
negative ion other than
hydroxide (OH⁻) Salts.

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