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ANSWER Answer
the following

questions in the
space provided. 1.

Write the
equilibrium
expression for the
following
hypothetical

equation: $3A(aq)$

$B(aq) \rightleftharpoons 2C(aq)$

$3D(aq)$ 2. a. Write

the appropriate

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expression for each
of the following
equations. Include
the value of K ...

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upset the
equilibrium of a
chemical system?
equilibrium
constant.

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ANSWER Answer

the following
questions in the
space provided. 1.
Match the solution
type on the right to
the corresponding

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Answers

relationship
between the ion
product and the
K_{sp} for that
solution, listed on
the left. _____ The
ion product
exceeds the K_{sp}.

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Answers
sometimes referred
to as the chemical-
equilibrium

expression. The H_2 , I_2 , HI

Equilibrium System
Consider the

reaction between H_2
and I_2 vapor in a
sealed flask at an
elevated

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temperature. The
rate of reaction can
be followed by

observing the rate
at which the violet
color of the iodine
vapor diminishes,

as $[C]^x[D]^y$

$[A]^n[B]^m$ 556

CHAPTER 18

CHAPTER 18

Chemical

Equilibrium

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Modern Chemistry
Answers

1. In a bottle of unopened cola, the CO_2 gas dissolved in the liquid is in equilibrium with the CO_2 gas above the liquid. The dissolved gas reacts with water molecules in the cola to form carbonic acid, which also dissociates into

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carbon dioxide and
water. Which
chemical
equation(s) best
describe this
equilibrium
system? </br> a.

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