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Review |
Thermochemistry Unit**

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Section 1: The Flow of Energy – Heat and Work. What is thermochemistry? The study of the energy changes that occur in changes of state or during chemical reactions.

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)

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CHEMISTRY

Chemistry at Scotch

Plains Fanwood Hs. Cha

pter 17 18 Thermochemist

ry, Reaction Rates, and Eq

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Chapter 17 review

Important Vocabulary

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Thermochemistry 429

Section Review

Objectives • Explain the relationship between energy, heat, and work •

Distinguish between exothermic and

endothermic processes •

Distinguish between heat capacity and specific heat

Vocabulary Key

Equations and

Page 9/30

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Relationships • 1

Calorie 1 kilocalorie

1000 calories • 1 J

0.2390 cal and 4.184 J 1

cal • C m

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Chapter 17:

Thermochemistry 17.1

The Flow of Energy –

Heat and Work * * *

Standard heats of

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formation are used to calculate the enthalpy change for the reaction of carbon monoxide and oxygen. Interpreting Diagrams How does this diagram also demonstrate Hess's law? * * * * *

Enthalpy changes accompany changes in state.

Science / Chapter 17 -

Page 11/30

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thermochemistry (handouts)

Ch 17 Thermochemistry

Practice Test Matching

Match each item with

the correct statement

below. a. calorimeter d.

enthalpy b. calorie e.

specific heat c. joule f.

heat capacity _____ 1.

quantity of heat needed

to raise the temperature

of 1 g of water by 1°C

_____ 2. SI unit of

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energy _____ 3.
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chapter explores ideas

related to heats of

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reaction. Students will be exploring endothermic and exothermic processes, phase changes and Hess's Law.

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section review answers

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Y WORKSHEET KEY

1. The following

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describes the reaction
that takes place when a
typical fat, glyceryl
trioleate, is metabolized
... 356 Chapter 6

Worksheet Keys

CHEMISTRY 151 - DH
AND DE

WORKSHEET KEY 1.

Nitrogen dioxide gas
reacts with liquid water
to yield liquid nitric acid
and nitrogen monoxide
gas. ... T g C 35.72

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17.02 C g C ...
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Chapter 10 - moles
(handouts) Chapter 11 -
reactions (handouts)
Chapter 12 -
stoichiometry
(handouts) Chapter 13 -
states of matter
(handouts) Chapter 17 -

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thermochemistry

(handouts) Chapter 18 -

reaction rates (handouts)

Chapter 19 - acids,

bases, and salts

(handouts) Material

Science Schedule.

Previous weeks

schedule; Chemistry

Basics ...

**THERMOCHEMIST
RY WORKSHEET**

KEY - Faculty

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Chapter 17 review Important Vocabulary •

Thermochemistry-

Studies of energy
changes that occur

during chemical
reactions and change

state • Chemical

potential energy- The
energy stored in the

chemical bonds of a
substance • Heat-

Represented by q , is
energy that transfers

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from one object to

another
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Chemistry 17.2

Thermochemistry Test

Preview Matching

Match each item with

the correct statement

below. a. calorimeter d.

enthalpy b. calorie e.

specific heat c. joule f.

heat capacity _____ 1.

quantity of heat needed

to raise the temperature

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of 1 g of water by 1 C

2. ___ SI unit of energy

3. ___ quantity of heat

needed to change the

temperature of 1 g of a

...

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chemistry

thermochemistry ... -

Quizlet

Chapter 17

Thermochemistry 187

10. Complete the

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enthalpy diagram for the
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combustion of natural
gas. Use the

thermochemical

equation in the first

paragraph on page 517

as a guide. SECTION

17.3 HEAT IN

CHANGES OF STATE

(pages 520–526) This

section explains heat

transfers that occur

during melting,

freezing, boiling, and

freezing, boiling, and

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condensing.
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**Thermochemistry Test
Preview**

This packet is used to help students study for the thermochemistry test. Answers are given for quantitative problems so that students may check their work. Purpose: To help student prepare for an exam. ...

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Practice Worksheet

Answer Key . Cheeto

Lab Activity .

McLaughlin, Kimberly

/ Thermochemistry

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chemistry chapter 17.

toluene. phenol. ...

Pearson Chemistry

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**Chapter 17-18 Packet -
,Reaction Rates, and Eq
uilibrium ...**

CHAPTER 17,

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(continued) 2. What is calorimetry?

Calorimetry is the accurate and precise measurement of heat change for chemical and physical processes. 3.

Use Figure 17.5 on page 511. Circle the letter next to each sentence that is true about calorimeters. G The calorimeter container is

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insulated to minimize
loss of heat to or

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Thermochemistry -

Mrs. Gingras'

Chemistry Page

IPOD #29 – Chapter 13

concepts Review –

Phase Changes Slide 24:

Describe terms for

phase changes

Worksheet: Chapter 17

– Thermochemistry

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Notes Chapter 17 Notes,
Packet, Answers

Slides 1-3: What is thermochemistry?

Chapter 17 Notes,

Slides 4-11: Heat, energy, calorie/joule conversions

Chapter 17

Notes, Slides 12-13:

Endo vs. Exo

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