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Central Science
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Additional
Aspects of
Aqueous

Equilibria Water
is important
because of its

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exceptional

ability to

dissolve wide

variety of

substances Two

additional types

of aqueous

equilibria o

Those involving

slightly soluble

salts in

solutions o

Those involving

the formation of

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*metal complexes
in solutions*
Chemistry

*Chapter 17 -
Additional
Aspects of
Aqueous
Equilibria: Part
1 of 21*

*In this lecture
I'll teach you
how to calculate
the pH of a
buffered*

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*solution using
both the common
ion effect
approach and the
Henderson-
Hasselbalch
equation. I'll
also teach you
what an acid ...*

*Chapter 17
Review Answers -
CHAPTER A
LIQUIDS Practice
Page 10/37*

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CHAPTER 17

EQUILIBRIUM: THE EXTENT OF CHEMICAL REACTIONS 17.1

*If the rate of
the forward
reaction exceeds
the rate of
reverse
reaction,
products are
formed faster*

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*than they are
consumed. The
change in
reaction
conditions
results in more
products and
less reactants.
A change in
reaction*

Chapters 17-18

Review -

Chapters 17

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Liquid Water &
Its ...

In this lecture
I'll teach you
how to about the
common ion
effect and how
to perform pH
calculations for
common ion
effect problems.
I'll also teach
you what
buffered

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solutions are
Chemistry
and how to ...

*Chemistry: The
Central Science
Chapter 17:
Additional ...
Chapters 17-18
Review -
Chapters 17
Liquid Water&
Its Properties
Chapter 18
Solutions Name
Page 14/37*

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Answers from
CHEMISTRY CHEM1B
at Laney
College. CHAPTER
A LIQUIDS
Practice Problem
page 428 . 1 I\
g. 1. Using
Figure 17.7, ?nd
boiling points
of 4. Using

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Chapter 17 -

Additional

Aspects of

Aqueous

Equilibria: Part

2 of 21

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that cover the

chemistry and

scientific

concepts you

need to learn or

review. ...

Prentice Hall

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...

Chapter 17

Page 19/37

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Review:

Thermodynamics .

*First we must
review material
from chapter 8.*

*You are
responsible for
knowing material
from first
semester,*

*specifically
sections 2, 5,
7, 9 and 10*

except where

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*Chemistry
Homework
Problems
Chemistry is the
study of the
composition,
properties, and
reactivity of
matter. This may
be your first
time taking*

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*chemistry, but
chances are you
know a lot
already from
observing the
world around
you. We will be
covering the
material in a
first year
introductory
high school or
college general
chemistry*

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Kinetics chapter

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learn the

essential

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*associated with
reaction
kinetics.*

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chemistry
concepts*

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Chapter 17

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Review Thermo -
Glendale
Community*

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General

Chemistry

Problems are
keyed to Kotz
and Treichel
Chemistry &
Chemical ...

Chapter 17, Acid-
Base Homework

Problem Set 1;

Chapter 18, Acid-
Base Homework

Problem Set 2

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*... Chapter 21,
Electrochemistry
Homework Problem
Set; Chapter 24,
Nuclear
Chemistry
Homework Problem
Set. Please send
comments or
suggestions to s
vanbram@science.
widener ...*

CHAPTER 17

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*EQUILIBRIUM: THE
EXTENT OF*

*CHEMICAL
REACTIONS*

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Chapter 17

Additional

Aspects of

Aqueous

Equilibria

You can test

Page 31/37

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*your readiness
to proceed by
answering the
Review Questions
at the end of
the chapter.*

*This might also
be a good time
to read the*

Chapter ... 658

*Chapter 17 An
Introduction to
Organic*

Chemistry,

Where To
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Biochemistry,
Chemistry, ... 660 Chapter

17 An

Introduction to
Organic

Chemistry,

Biochemistry,

and Synthetic

Polymers. 17.1

Organic

Compounds 661

Holt McDougal

Modern Chemistry

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Chapter 17:

Reaction ...

This video explains the concepts from your packet on Chapter 17 (Additional Aspects of Aqueous Equilibria), which can be found here: <http://goo.gl/iV95W>

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synthetIC p
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