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correctly. In  
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terms at the

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16 — The Breast  
Answers: Chapter  
16 Matching 1. f 2.  
c 3. h 4. k 5. j 6. n  
7. b 8. g 9. l 10. d  
11. m 12. e 13. a  
14. i Image

Labeling 1A.  
Pectoralis major  
muscle 1B. Skin  
1C. Subcutaneous  
fat layer 1D. Deep

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fascial layer 1E.  
Superficial fascial  
layer 1F.

Lactiferous duct  
1G. Lactiferous  
sinus 1H. Nipple 1I.  
Areola 1J.  
Parenchymal layer  
1K ...

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Worksheet page 2

Table 3. 3D

Perspective of  
River On the figure  
above, draw your  
proposed locations  
for agricultural  
fields (draw a small  
square) and five  
possible homesites  
for a house (use a  
circle or simple

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house symbol).

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Section Review

Objectives •

Identify the factors  
that determine the  
rate at which a  
solute dissolves •

Identify the units  
usually used to

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express the solubility of a solute • Calculate the solubility of a gas in a liquid under various pressure conditions • Identify the factors that determine the mass of solute that will dissolve in a given

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**Solved: Chapter  
16 Investigation  
Worksheet Page  
2 Table 3 ...**

the answer. 10 19  
105 10 14; the  
answer will be  
about 20 10 14, or  
2 10 13. c.

Calculate your  
answer. Check it  
against your  
estimate from part  
b. 1.7 10 13 kg



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m/s<sup>2</sup> d. Justify the number of significant digits in your answer. The least-precise value is 4.5 T, with 2 significant digits, so the answer is rounded to 2 significant digits.

16.

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questions in the  
spaces provided. If

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you run  
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**Answers:**

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Chapter 16

Solutions 167

SECTION 16.1

PROPERTIES OF

SOLUTIONS (pages  
471-477) This

section identifies  
the factors that

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Assess

affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471–472) Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2.

1.

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**PHYSICS 202**

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About This Quiz & Worksheet. Certain ingredients mix well together while others do not. This quiz/worksheet combo will help test your understanding of the different types

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of mixtures, such  
as ...

Answers

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Properties of

Solutions 16.2

Concentrations of

Solutions 16.3

Colligative

Properties of

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solubility and supersaturated solutions to explain your answer.

Crystal-growing kits usually begin with a

supersaturated solution. When a seed crystal is added to the solution, crystals rapidly begin ...

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For ANY solution, if you know the concentration of either the hydrogen ion or the hydroxide ion then you know the concentration of the other! They are

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ALWAYS related by  
the equation  
above!  $\text{pH} =$   
 $-\log[\text{H}_3\text{O}^+] \dots$   
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Solutions I.  
Solutions A.

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Solution is a homogeneous mixture involving two or more pure substances. Its composition usually can be varied within certain limits. B. Solute substance dissolved in the solution. C. Solvent the substance in which the solute is

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Answer  
dissolved Example:  
Salt + H<sub>2</sub>O H<sub>2</sub>O is  
the solvent NaCl  
Salt is the solute  
Na<sup>+</sup>Cl<sup>-</sup> II.

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Answers: (a) basic,  
(b) neutral, (c)  
acidic. Acids and  
Bases ... A student  
prepared a 0.10 M  
solution of formic  
acid ( $\text{HCHO}_2$ ) and  
measured its pH  
using a pH meter.  
The pH at 25 °C  
was found to be  
2.38. (a) Calculate

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Base Equilibria and  
Solubility Equilibria

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Worksheet I A table

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of ionization constants and  $K_a$ 's is required to work some of the problems in this chapter. 1. In which one of the following solutions will acetic acid have the greatest percent ionization?

**05 CTR ch16**

**7/12/04 8:14 AM**

*Page 35/38*

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they can have the  
ability to utilize  
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effectively on his or  
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