

Chapter 15 Water Aqueous Systems Worksheet Answers

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Chapter 15 (Water and Aqueous Systems) Test Study Guide ? The bonds between the hydrogen and oxygen atoms in a water molecule are polar covalent bonds. ? The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. ? The bonds between adjacent water molecules are called hydrogen bonds.

Chapter 15 - Water and Aqueous Systems - 15 Assessment ...
The high surface tension of water is due to the: a. small size of water molecules. b. low mass of water molecules. c. hydrogen bonding between water molecules. d. covalent bonds in water molecules. _____ 12.
Salts and other compounds that remove moisture from air are said to be: a. efflorescent. c. colloidal. b. surfactant. d. hygroscopic. 10 9 ...

Prentice Hall Chemistry Chapter 15: Water and Aqueous ...
Chapter 15 - Water and Aqueous Systems - 15 Assessment - Page 511: 64 Answer Anions are attracted to the hydrogen atoms of the water molecule, because hydrogen atoms are slightly positive.

Chapter 15 - Water and Aqueous Systems - 15.1 Water and ...
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Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous ...
Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its Properties - 15.1 Lesson Check - Page 493: 1
Answer The hydrogen bonding in water cause the high surface tension and low vapor pressure of water.

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WATER AND AQUEOUS SYSTEMS chapter ... Crystals of copper sulfate pentahydrate always contain five molecules of water for each copper and sulfate ion pair.

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05 CTR ch15 7/12/04 8:14 AM Page 387 WATER AND AQUEOUS ...
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19 - Acids, Bases and Salts Chapter 20 - Oxidation-Reduction Reactions

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1. Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and its Properties OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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Chapter 15 Water and Aqueous Systems 159 SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure

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Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous Aqueous Systems - 15.3 Lesson Check - Page 507: 21 Answer Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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