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Equilibrium cannot be established between the liquid and the gas phase if the top is removed from the bottle because the system is not closed; one of the components of the equilibrium, the Br_2 vapor, would escape from the bottle until all liquid disappeared.

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Thus, more liquid would evaporate than can condense back from the gas phase to the liquid phase.

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Chapter 13 Study Guide Chapter 13 Section 1
salinity: amount of salt dissolved in water
thermocline: middle layer in ocean, does not receive a lot of sunlight so temperature drops drastically
photosynthesis: process which plants make food, they use CO₂ and sunlight to produce a type of sugar
density: when two items take up the same amount of ...

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10. volt Section 13.3 11. ohm 12. Ohm's law
13. resistance 14. potentiometer 15.
conductor Reviewing Concepts Section 13.1 1.

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reserves make a population experience delayed
density dependence? a. They enable the
population to survive above carrying

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capacity. b. They protect organisms from devastating events like hurricanes or floods. c.

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