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workplace.**

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**Review Solutions
Section 2**

**Answers. Modern
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CHAPTER 1

**REVIEW. Matter
and Change.**

SECTION 1.

SHORT ANSWER

Answer the

12. 18. B. 18. 12.

6. A. MODERN

CHEMISTRY.

MEASUREMENTS

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REVIEW "'--
Solutions .
SHORT ANSWER
Answer the

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**following
questions in the
space provided.**

- 1. Describe the errors made by the following students in making molar solutions. 3.**
- James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl**

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and adds 1 L of
water to the
solid. James
made several
errors.

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terms, and more
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Chemistry 1

Solutions

CHAPTER 12

REVIEW

Solutions

Teacher Notes

and Answers

Chapter 12

SECTION 1

SHORT ANSWER

1. c 2. a 3. b 2. a.

alcohol b. water

c. the gels 3. The

mixture is a

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colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

CHAPTER 12

REVIEW

Solutions -

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**Manasquan
Public Schools**

**The Solutions
chapter of this
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**Chemistry
Companion
Course helps
students learn
the essential
lessons**

**associated with
solutions.**

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**CHAPTER 12
REVIEW**

**Solutions -
Weebly
Modern
Chemistry
Chapter
12-Solutions
Section 1- Types
of Mixtures
Solutions are
homogeneous
mixtures of two**

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or more substances in a single phase.

Soluble describes a substance as capable of being dissolved.

Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved

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in a solution.
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ICSE, both the

boards

recommend and

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CHAPTER 12
REVIEW
Solutions
SECTION 2
SHORT ANSWER**

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Solutions, Modern

Chemistry

Answer the
following
questions in the
space provided.

- 1. The following are statements about the dissolving process. Explain each one at the molecular level.**
 - a. Increasing the pressure of a solute gas above**

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Solutions Modern

**a liquid solution
increases the**

**solubility of the
gas**

Modern

Chemistry

Chapter 12

Solutions Review

Answers

Modern

Chemistry 107

Solutions

CHAPTER 12

Page 24/46

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REVIEW
Solutions

SECTION 3

SHORT ANSWER

Answer the following questions in the space provided.

1. Describe the errors made by the following students in making molar solutions. a.

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**James needs a
0.600 M solution
of KCl. He
measures out
0.600 g of KCl
and adds 1 L of
water to the
solid.**

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Review CHAPTER

12 & 13:

SOLUTIONS The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test.

Focus on this content, but

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**make sure to
review class**

**notes, activities,
handouts,**

questions, etc. If

you study this

document and

NOTHING else,

you should at

least be able to

PASS the test.

Chapter 12 and

13 REVIEW.docx -

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**Chapter 12 and
13 Review Works
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Term Description

12.3-13.3 424,

449 Charles

Lindbergh Made

the first nonstop

solo flight across

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**the Atlantic
Ocean in his
plane the Spirit
of St. Louis. He
became a hero
upon his return
home, receiving
a ticker tap
parade and a
reception form
the white house.**

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**Chapter
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**Section 1- Types
of Mixtures**

**Solutions are
homogeneous
mixtures**

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Modern

Chemistry

Chapter

12-Solutions

Section 1- Types

Page 33/46

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Chapter 12 Review

Solutions Modern

Chemistry

of Mixtures

Solutions are

homogeneous

mixtures of two

or more

substances in a

single phase.

Soluble

describes a

substance as

capable of being

dissolved.

Solvent is the

dissolving

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Solutions, Modern
Chemistry
medium in a
solution. Solute
is the substance
that is dissolved
in a solution.

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**CHAPTER 12:
SOLUTIONS**

**Sections 12.1 -
12.4 (Tro, 2 nd
edition) 1.**

**Terminologies:
solution, solute,
solvent, miscible
(vs. immiscible)
liquids,
dissolution, 2.**

**Know the
common types of
solutions and**

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give examples of each (Table 12.1)

3. Concepts: Be able to explain the following: (a) Intermolecular forces (IMFs) of attractions that exist in solution and relative

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Chapter 12**

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TEACHER.

Chemistry Ch. 10

Vocab. 37 Terms.

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Solutions PART I
In the space

provided, write
the letter of the
term or phrase
that best ...

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Chapter Test

Name Class Date

Chapter Test B,

continued _____ 7.

... 12. 760 mm

13. V P VP 14.

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effusion 15.

pressure 16.

newton 17.

barometer 18.

partial pressure

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12-Solutions Section 1- Types of Mixtures

Solutions are homogeneous mixtures of two or more substances in a single phase.

Soluble describes a substance as capable of being dissolved.

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Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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