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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 12 71 CHAPTER 12 STUDY GUIDE FOR CONTENT MASTERY Section 12.3 Limiting Reactants In your textbook, read about why reactions stop and how to determine the limiting reactant. Study the diagram showing a chemical reaction and the chemical equation that repre-sents the reaction.

Chapter 12 Content Mastery Stoichiometry

STUDY GUIDE FOR CONTENT MASTERY Section 12.2 Stoichiometric Calculations In your textbook, read about mole-to-mole conversion. ... Chapter 12 Stoichiometry . In the reaction represented by the equation $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$, how many grams of hydrogen are produced if 120. g of Na and 80.0 g

Answer Key - Chemistry 2014-2015

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12. 13. 14. 72 Methanol Oxygen gas Carbon dioxide Water What are the reactants? C) What are the products? 10B What is the total mass of the reactants? What is the total mass of the products? How do the total masses of the reactants and products compare? Chemistry: Matter and Change Chapter 11

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