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10.E: Nuclear and
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questions covering
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10 Prep-Test**
Nuclear Chemistry is
quite an interesting field
of study, both for the
fact that it is scary and
informative at the same

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time. Find out how much you know. If you are prepared to test your knowledge try it out.

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Section 10.1

Radioactivity (pages
292–297) Quick Upload

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292 Chapter 10 A B 292

Chapter 10 FOCUS

Objectives 10.1.1

Describe the process of
nuclear decay. 10.1.2

Classify nuclear
radiation as alpha
particles, beta particles,

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or gamma rays. 10.1.3

Balance nuclear
equations. 10.1.4

Identify sources of
nuclear radiation, and
describe how nuclear
radiation affects matter.

10.1.5 Describe
methods of ...

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A comprehensive
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Chemistry ...

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10 Nuclear Chemistry
Solutions to In-Chapter
Problems 10.1 Refer to
Example 10.1 to answer
the question. The
atomic number (Z) = the
number of protons. The
mass number (A) = the
number of protons + the
number of neutrons.
Isotopes are written with
the mass number to the

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upper left of the element
symbol and the

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Decay A wooden object
from an archeological
site is subjected to
radiocarbon dating. The
activity of the sample
that is due to ^{14}C is

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measured to be 11.6
disintegrations per
second.

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PS Week 03. PS WEEK
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Nuclear Chemistry.

Artificial

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unstable atomic nucleus

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Radioisotope. Any atom containing an unstable nucleus. What happens during nuclear decay? Atoms of one element can change into atoms of a different element together.

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Who discovered radioactivity and in what year? Henri Becquerel in 1896

radioactivity the process in which an unstable atomic nucleus emits charged particles and energy any atom containing an unstable nucleus is a

**RADIOACTIVE
ISOTOPE or
RADIOISOTOPE**

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nuclear decay when
atoms of one element
can change into atoms
of a completely different
element nuclear
radiation [□]

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Radioactivity ...
CHEMISTRY: Chapter
10 Prep-Test Matching
Match each item with
the correct statement

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below. a. calorimeter d.
temperature b. calorie e.
specific heat c. joule f.
heat ____ 1. quantity of
heat needed to raise the
temperature of 1 g of
water by 1 C ____ 2. SI
unit of energy ____ 3.
energy transferred
between 2 objects
because of temperature
difference

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