

Ch 22 Hydrocarbon Compounds Workbook Answers

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This section describes the bonding in hydrocarbons and ...

22.4 Hydrocarbon Rings • Some hydrocarbon compounds have a carbon chain that is in the form of a ring. • In a benzene molecule, the bonding electrons between carbon atoms are shared evenly around the ring. 22.5

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Hydrocarbons From Earth's Crust • Natural gas is an important source of alkanes of low molar mass.

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***CHAPTER 22, Hydrocarbon Compounds
(continued) SECTION 22.4 HYDROCARBON
RINGS (pages 709-711) This section describes
how to identify and classify cyclic
hydrocarbons. It also explains the bonding in
benzene. Cyclic Hydrocarbons (page 709) 1.
What is a cyclic hydrocarbon? It is a compound
that contains a hydrocarbon ring. five 2.***

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CHAPTER 22, Hydrocarbon

Compounds(continued) 14. Circle the letter of each condensed structural formula for pentane.

a. C 5 H 12 b. CH 3 CH 2 CH 2 CH 2 CH 3 c. CH 3 (CH 2) 3 CH 3 d. C — C — C — C — C

15. The IUPAC system uses _____ to show the number of carbon atoms in a straight-chain alkane. 16.

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at the end of the chapter. Workbook If your instructor recommends the Active Learning Workbook, do Questions, Exercises, and Problems 1-18. Chapter 22-Assignment B: More Hydrocarbons: Alkenes, Alkynes, and Aromatics The alkanes of Assignment A are called saturated because they can form no additional bonds.

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Chapter 22 - Hydrocarbon Compounds. STUDY. PLAY. hydrocarbon. an organic compound that contains only carbon and hydrogen. alkane. a

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hydrocarbon containing only single covalent bonds; saturated hydrocarbons. straight-chain alkane. a saturated hydrocarbon that contains any number of carbon atoms arranged one after another in a chain.

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- 22.1 Lesson Check - Page 771 9 including work
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Chapter 22

The Hydrocarbon Compounds chapter of this

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Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with hydrocarbon compounds.

***SECTION 22.1 HYDROCARBONS(pages 693-701)
Compounds with the same molecular formula but the atoms are joined together in a different order Differ in physical properties (e.g., boiling point, melting point) and chemical reactivity
Branched-chain alkanes tend to have lower boiling points than straight-chain (does not together as well)***

SECTION 22.1 HYDROCARBONS(pages 693-701)

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Terms in this set (23) Alkanes. a hydrocarbon
containing only single covalent bonds. Alkenes.***

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Hydrocarbons with one or more carbon-carbon double bonds. alkyl group. a hydrocarbon substituent.

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744 Chapter 21 • Hydrocarbons Section

221.11.1 Introduction to Hydrocarbons MAIN

Idea Hydrocarbons are carbon-containing organic compounds that provide a source of energy and raw materials. Real-World Reading Link If you have ridden in a car or a bus, you have used

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