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Notes Chemistry Chapter 7 I

Group 15 Elements: The Nitrogen

Family 1. In group 15 of the

Periodic Table, the elements,

nitrogen (7N), phosphorus (15P),

arsenic (33As), antimony (51Sb)

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and bismuth (83Bi) are present.
2. (a) The elements of this group can exhibit various oxidation states ranging between -3 to+ [...]

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Chemistry. NCERT IN TEXT

QUESTIONS 7.1. Why are

pentahalides more covalent than

trihalides? Ans: The group 15

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elements have 5 e-1 s in their valence shell. It is difficult to lose 3e-1 s to form E 3+ and even more difficult to lose 5e-1 s to form E 5+. Thus, they have very little tendency to form ionic compounds.

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