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The p-Block Elements Class 12 Notes

Chemistry Chapter 7 | Group 15

Elements: The Nitrogen Family 1. In group 15 of the Periodic Table, the elements, nitrogen (7N), phosphorus (15P), arsenic (33As), antimony (51Sb) and bismuth (83Bi) are present. 2. (a)

The elements of this group can exhibit various oxidation states ranging between -3 to+ [...]

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IN TEXT QUESTIONS 7.1. Why are
pentahalides more covalent than
trihalides? Ans: The group 15 elements
have 5 e⁻ s in their valence shell. It is
difficult to lose 3e⁻ s to form E³⁺ and
even more difficult to lose 5e⁻ s to form
E⁵⁺. Thus, they have very little tendency
to form ionic compounds.

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