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Lithium - Wikipedia

Nuclear chemistry is the sub-field of chemistry dealing with radioactivity, nuclear processes, and transformations in the nuclei of atoms, such as nuclear transmutation and nuclear properties.. It is the chemistry of radioactive elements such as the actinides, radium and radon together with the chemistry associated with equipment (such as nuclear reactors) which are designed to perform nuclear ...

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Lithium (from Greek: ??????, romanized: lithos, lit. 'stone') is a chemical element with the symbol Li and atomic number 3. It is a soft, silvery-white alkali metal. Under standard conditions, it is the lightest metal and the lightest solid element. Like all alkali metals, lithium is highly reactive and flammable, and must be stored in

vacuum, inert atmosphere or inert liquid such as ...

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*R.C. Ropp, in Encyclopedia of the Alkaline Earth Compounds, 2013
Calcium Chloride. Calcium chloride, CaCl_2 , is a typical ionic halide
and is a solid at room temperature. Its molecular weight is 110.98
g/mol and its melting point is 772 °C. Very few natural minerals occur.
The occurrence of a dihydrate (mineral "Sinjarite") and hexahydrate
("Antarcticite") is very rare and is connected ...*

Nuclear chemistry - Wikipedia

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