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4. The electrolysis of copper(II) sulfate solution. The electrolyte copper(II) sulfate, provides a high concentration of copper(II) ions Cu^{2+} and sulfate ions SO_4^{2-} to carry

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the current during the electrolysis process. There are tiny concentrations of hydrogen ions H^+ and hydroxide ions (OH^-) from the self-ionisation of water itself, but these can be ignored in this experiment.

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