

## Answer Key For Counting Atoms In Compound

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CLASSWORK ~ Practice Counting Atoms Worksheet . Directions for each problem . 1) write down the different elements in each compound. 2) write down how many of that particular atom there are. 3) how many atoms are there total in the compound. Examples: A)  $\text{MgCl}_2$  Mg - 1 B) 5  $\text{ZnSO}_4$  Zn - 5

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Counting Atoms The formula for a compound indicates the elements that make up the compound and the number of atoms of each element present in the compound. These numbers of atoms are indicated by the use of

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small numbers called subscripts. Sometimes groups of atoms act as a single atom. Such a group of atoms is called a polyatomic ion.

### **How to Count Atoms**

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the amount of a substance that contains as many particles as there are atoms in exactly 12 g of carbon-12 Avogadro's number  $6.022 \times 10^{23}$  is the number of particles in exactly one mole of a pure substance,  $6.022 \times 10^{23}$

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