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In this experiment, you will determine the  
molar volume of a gas by conducting a

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chemical reaction that produces a gas, as shown in the reaction equation below. You will react a known mass of solid magnesium with an excess of hydrochloric acid, in a sealed vessel, and use the pressure change to ...

The Molar Volume of a Gas | Experiment #5 from Advanced ...

CHEM-A #10: In this experiment, you will Prepare and test standard solutions of  $\text{FeSCN}^{2+}$  in equilibrium. Test solutions of  $\text{SCN}^-$  of unknown molar concentration. Determine the molar concentrations of the ions present in an equilibrium system. Determine the value of the equilibrium constant,  $K_{eq}$ , for the reaction.

The Determination of an Equilibrium Constant - Vernier  
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process used to determine the volume of a  
solution that is needed to react with a  
given amount of another substance. In  
this experiment, your goal is to determine  
the molar concentration of an acid  
solution by conducting titrations with a  
base of known concentration.

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