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called capillary action.

Figure $\backslash(\backslash\text{PageIndex}\{1\}\backslash)$:

Adhesion: Capillary action in a glass tube is caused by the adhesive forces exerted by the internal surface of the glass exceeding the cohesive forces between the water molecules themselves. Why are cohesive and adhesive forces important for life?

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fundamentals of each of these areas, as they are important for the practitioner of this joining technology.

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Adhesion is a phenomenon that takes place at the interfaces of adherends and adhesives. An understanding of the forces that develop at the interfaces is essential in the selection of the right adhesive, proper surface treatment of adherends, and effective and economical processes to form bonds.

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